

Lesson 5 - Metals and Non-Metals

71 Lu Lutetium	99 Es Einsteinium	18 Ar Argon	7 N Nitrogen	53 I Iodine	7 N Nitrogen	31 Ga Gallium			
53 I Iodine	7 N Nitrogen	90 Th Thorium	68 Er Erbium	7 N Nitrogen	69 Tm Thulium	53 I Iodine	76 Os Osmium	7 N Nitrogen	16 S Sulfur

By the end of this lesson I will be able to

- ★ describe where on the periodic table metals and non-metals are.
- ★ can comment on the appearance of metals
- ★ link the uses of metals to their properties

Task 1: Element Classification

Element	Group
Lithium	
Chlorine	
Argon	
Neon	
Astatine	
Francium	
Bromine	

Elements can be categorised in a number of ways, we have so far learned about some of the groups that metals can be categorised into.

Examples of these groups are:

1. _____
2. _____
3. _____

Elements are also categorised due to their properties.

You are now going to categorise elements as metals and non-metals and look at their properties.

Task 2 - Metals & Non-Metals

Elements can also be categorised as _____ or _____ .

The elements of the periodic table are split into metals on the _____ (under the magic staircase) and non-metals on the _____ (above the magic staircase).

Draw the magic staircase onto the periodic table below and colour it to show the metals and non-metals.

hydrogen 1 H 1.0079																				helium 2 He 4.0026
lithium 3 Li 6.941	beryllium 4 Be 9.0122													boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180	
sodium 11 Na 22.990	magnesium 12 Mg 24.305													aluminium 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948	
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80			
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29			
caesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 * Lu 174.97	lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]		
francium 87 Fr [223]	radium 88 Ra [226]	89-102 * * Lr [262]	lanthanum 103 La [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	unnilium 110 Uun [271]	ununium 111 Uuu [272]	unbibium 112 Uub [271]	ununquadium 114 Uuq [289]							

* Lanthanide series

lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.04
---------------------------------	------------------------------	------------------------------------	---------------------------------	---------------------------------	--------------------------------	--------------------------------	----------------------------------	-------------------------------	----------------------------------	-------------------------------	------------------------------	-------------------------------	---------------------------------

** Actinide series

actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]
-------------------------------	-------------------------------	------------------------------------	------------------------------	--------------------------------	--------------------------------	--------------------------------	-----------------------------	--------------------------------	----------------------------------	----------------------------------	-------------------------------	-----------------------------------	--------------------------------

Metals

Non-Metals

Task 3: Metal or Non-Metal

Use the periodic table to help you identify if the elements in the table are metal or non-metal.

Element	Metal/non-metal
copper	
sulfur	
tin	
bromine	
chlorine	
lead	
aluminium	
hydrogen	
iron	
carbon	
carbon - graphite	

Task 4: Conductor or Non-Conductor

_____ will conduct an electrical current.

_____ will not conduct an electrical current.

The exception to this rule is _____ which is a _____ and will conduct electricity.

Below is the method for an experiment to test conductivity of elements.

1. Collect all equipment
2. Connect the wires to the bulb. Place a crocodile clip at the other end of each wire.
3. Connect the crocodile clips to the first sample to be tested.
4. Record if the bulb lit up or not in the results table.
5. Repeat steps 3-4 for the remaining 9 elements.

Decide whether the following elements would light the bulb in the experiment.

Element	Metal/non-metal	Bulb Light? (Y/N)
copper		
sulfur		
tin		
bromine		
chlorine		
lead		
aluminium		
hydrogen		
iron		
carbon		
carbon - graphite		

Was I successful?	Red	Yellow	Green
I can describe where on the periodic table metals and non-metals are.			
I can comment on the appearance of metals.			
I can link the uses of metals to their properties			

